

PHY2201 Summary Sheet 9

Ensemble of N gas atoms. Outer electron can reside in a “ground-state” energy level, or in an excited state, 1 eV above this. At 1000 K, what fraction of atoms lie in the excited state, relative to the ground-state?

Boltzmann distribution:

$$\frac{n_i}{N} = \frac{\exp(-\epsilon_i/k_B T)}{\sum_{i=0}^{\infty} \exp(-\epsilon_i/k_B T)}$$

n_i is no. of systems occupying a state of energy ϵ_i , when ensemble of N such systems is in thermal equilibrium at temp T . For a 2-level system, energies ϵ_1, ϵ_2 , relative occupancy of these levels is given by

$$\frac{n_2}{n_1} = \frac{\exp(-\epsilon_2/k_B T)}{\sum_{i=0}^{\infty} \exp(-\epsilon_i/k_B T)} \frac{\sum_{i=0}^{\infty} \exp(-\epsilon_i/k_B T)}{\exp(-\epsilon_1/k_B T)} = \exp(-(\epsilon_2 - \epsilon_1)/k_B T)$$

$$\boxed{\frac{n_2}{n_1} = \exp(-\Delta\epsilon/k_B T)}$$

with $\Delta\epsilon = \epsilon_2 - \epsilon_1$

Useful “rule of thumb”:

At room temperature (300K) the thermal energy $k_B T$ is 25 meV

Hence at 1000 K the thermal energy is $25 \text{ meV} \times 1000/300$.

$$\therefore n_2/n_1 = \exp(-1/(25 \times 10^{-3}/3)) = \exp(-12) = 5 \times 10^{-6}$$

$$\underline{n_2/n_1 = 6 \times 10^{-6}}$$

Cool the gas to 300 K:-

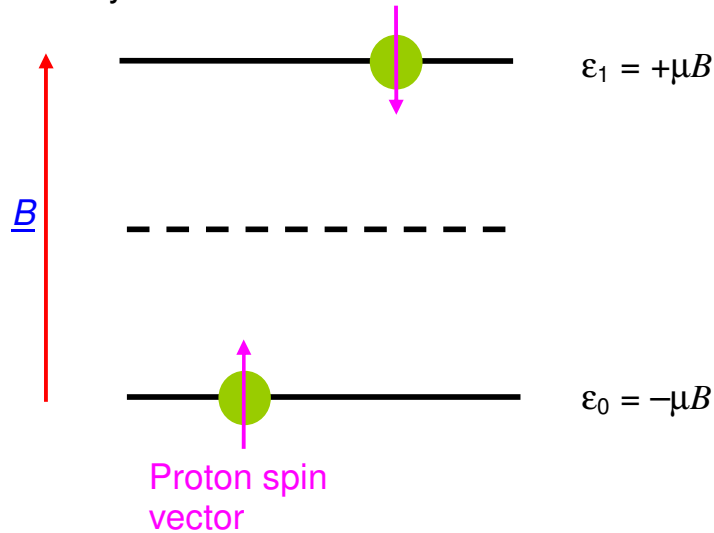
$$n_2/n_1 = \exp(-1/(25 \times 10^{-3})) = \exp(-40) = 5 \times 10^{-9}$$

$$\underline{n_2/n_1 = 4 \times 10^{-18}}$$

very strong T -dependence.

Ensemble of protons, magnetic moment μ , in external magnetic field B . Magnetostatic potential energy = $+\mu B$ if proton spin anti-parallel to field, $-\mu B$ if proton spin parallel to field.

Simple 2-level system:



In equilibrium, what is the net imbalance between spin “aligned” (n_{\uparrow}) and spin “anti-aligned” (n_{\downarrow}) protons (i.e. what is the fractional magnetization) at room temperature?

$$\frac{n_{\uparrow} - n_{\downarrow}}{n_{\uparrow} + n_{\downarrow}} = \frac{1 - n_{\downarrow}/n_{\uparrow}}{1 + n_{\downarrow}/n_{\uparrow}}$$

$$= \frac{1 - \exp(-(\epsilon_1 - \epsilon_0)/(k_B T))}{1 + \exp(-(\epsilon_1 - \epsilon_0)/(k_B T))} = \frac{1 - \exp(-2\mu B/(k_B T))}{1 + \exp(-2\mu B/(k_B T))}$$

Proton $\mu = 1.46 \times 10^{-26} \text{ JT}^{-1}$. $B = 1 \text{ T}$ (typically), $T = 300 \text{ K}$.

Hence $2\mu B = 2.9 \times 10^{-26} \text{ J}$. $k_B T = 1.38 \times 10^{-23} \times 300 \text{ J} = 3.12 \times 10^{-21} \text{ J}$.

Since $2\mu B \ll k_B T$ $\exp(-2\mu B/(k_B T)) \cong 1 - 2\mu B/(k_B T)$

$$\frac{n_{\uparrow} - n_{\downarrow}}{n_{\uparrow} + n_{\downarrow}} \cong \frac{1 - (1 - 2\mu B/(k_B T))}{1 + (1 - 2\mu B/(k_B T))} \cong \frac{\mu B}{k_B T}$$

Hence at 300K and 1 T, the net imbalance of proton spins is

$$1.46 \times 10^{-26} / 3.12 \times 10^{-21} \text{ J}$$

$$= 4.7 \times 10^{-6}$$

i.e. a very small imbalance.