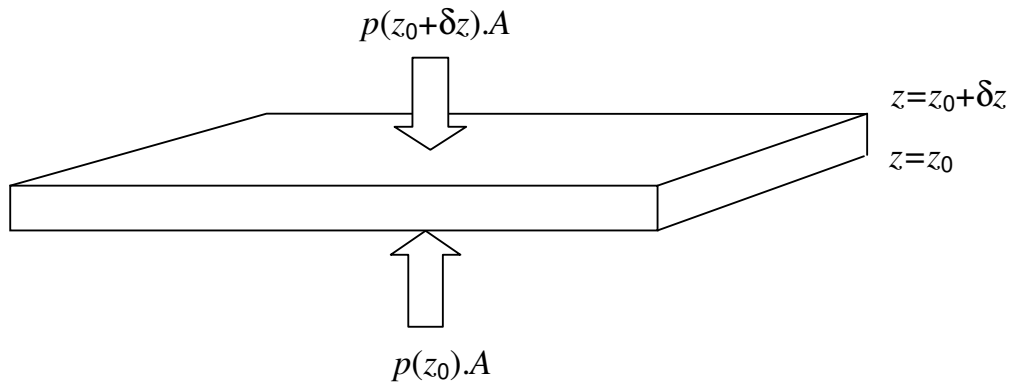


Example of the Boltzmann factor:

Consider a mass of isothermal ideal gas, at temperature  $T$ . For a thin slab of gas at height  $z_0$ , thickness  $\delta z$  and cross-sectional area  $A$  to not fall under gravity requires :



Hydrostatic equilibrium:-

$$p(z_0)A - p(z_0 + \delta z)A = \rho(z_0)A \delta z g$$

$$\therefore p(z_0) - p(z_0 + \delta z) = \rho(z_0) \delta z g$$

Expanding  $p(z)$  in a Taylor series:

$$p(z_0 + \delta z) = p(z_0) + \left. \frac{dp(z)}{dz} \right|_{z_0} \delta z + \frac{1}{2!} \left. \frac{d^2 p(z)}{dz^2} \right|_{z_0} \delta z^2 + \frac{1}{3!} \left. \frac{d^3 p(z)}{dz^3} \right|_{z_0} \delta z^3 + \dots O(\delta z^4)$$

In the limit that the slab thickness  $\delta z \rightarrow 0$ ,

$$p(z_0 + dz) - p(z_0) = \left. \frac{dp(z)}{dz} \right|_{z_0} dz ,$$

Hence,

$$\rho(z_0)g dz = - \left. \frac{dp(z)}{dz} \right|_{z_0} dz ,$$

Hence

$$\rho(z)g = -\frac{dp(z)}{dz}$$

$$\therefore \frac{dp(z)}{dz} = -n(z)m_A g$$

With  $m_A$  being the mass of one gas atom and  $n$  being the number density of gas atoms.

Ideal gas equation of state  $pV = Nk_B T \Rightarrow p = \frac{N}{V}k_B T = nk_B T$ , hence

$$k_B T \frac{dn(z)}{dz} = -n(z)m_A g \Rightarrow \frac{dn(z)}{dz} = -\frac{m_A g}{k_B T} n(z)$$

$$\therefore n(z) = n(z=0) \cdot \exp\left(-\frac{m_A g}{k_B T} z\right)$$

Hence  $n(z)$ ,  $\rho(z)$  and  $p(z)$  all fall exponentially with height.

$m_A g z$  is the gravitational potential energy of a gas atom at height  $z$ . Since  $n(z) \propto$  probability of finding a gas atom at height  $z$ , suggests that the probability of finding a gas atom in an “energy level” of value  $\epsilon(z)$  is proportional to

$$\exp\left(-\frac{\epsilon(z)}{k_B T}\right)$$

Boltzmann factor

This factor is of universal validity; whenever an ensemble of classical particles are in equilibrium at temperature  $T$ , the probability of an energy level of value  $\epsilon(z)$  being “occupied” by a particle of the ensemble varies as the exponential of  $-\epsilon(z)/k_B T$ .

Probability of a particle  
possessing this energy

